

SESQUITERPENOIDS RELATED TO JUVABIONE IN *ABIES PINSAPO*

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Abstract—From the acid fraction of the hexane extract of the wood of *Abies pinsapo*, four new sesquiterpenoids related to juvabione have been isolated: epitodomatuiic acid, *cis*-dihydroepitodomatuiic acid, 4'-dehydroepitodomatuiic acid and 3'-dihydroepitodomatuiic acid

INTRODUCTION

Abies pinsapo Boiss. is a tree which grows naturally on the high mountains at the western end of the Betic cordillera (Spain) and on the calcareous chain of Yebala (Northern Morocco). A superficial study of the polar extracts (acetone and diethyl ether) of the heartwood of *A. pinsapo* and its *marocana* subspecies was described in a previous paper [1]. Sitosterol, pinitol and secoisolariciresinol were identified. The components of the wood as well as leaves, bark, seeds and resin [2-6] have been studied in other species of *Abies*. Bowers *et al.* [7] isolated a substance from *A. balsamea*, that they called juvabione, because it exhibited high insect juvenile hormone activity when tested on the bug *Pyrrhocoris apterus* L. Juvabione and other related substances, which are of phytochemical interest, usually occur in the plant as methyl esters. The assigned stereochemistries are in general wrong [8-10] as shown by the use of the DC and DOR techniques [11].

RESULTS AND DISCUSSION

The acid fraction of the hexane extract from the wood of *A. pinsapo* was treated with diazomethane and four methyl esters (1-4) were isolated by column chromatography on silica gel. Upon saponification with methanolic potassium hydroxide these compounds yielded the corresponding natural sesquiterpene acids 5-8.

Saponification of the least polar ester (1) gave a substance with the molecular formula $C_{15}H_{24}O_3$ ($[M]^{+}$, $[M-H_2O]^{+}$ at m/z 234). Its IR spectrum showed absorption bands due to a ketone carbonyl and an α,β -unsaturated acid (1710 and 1686 cm^{-1} , respectively), as well as a trisubstituted C-C double bond (1646 cm^{-1}). Its $^1\text{H NMR}$ spectrum (Table 1) contained signals of a methyl group bound to a methine ($\delta 0.84$, d) and two methyls of an isopropyl group ($\delta 0.86$, d). At $\delta 7.10$, a multiplet due to the olefinic β proton of an α,β -unsaturated acid was observed. The acid proton appeared as a broad signal at $\delta 10.00$. These data together with the $^{13}\text{C NMR}$ data (Table 2) were consistent with a structure of a carboxylic acid related to juvabione. The positive sign of the optical rotation and the values of the chemical shifts in the $^{13}\text{C NMR}$ spectrum allowed us to establish the stereochemistry of C-4 and C-1'. Thus the δ values for C-3 and C-5 (28.8 and 26.1 , respectively) were

in agreement with the reported data [12] for an *S* configuration at C-1' of epijuabione (1). Therefore, the acid has structure 5, namely epitodomatuiic acid.

The substance formed on saponification of 2 showed in its mass spectrum a peak at m/z 254 ($[M]^{+}$, $C_{15}H_{26}O_3$), the other spectroscopic properties being very similar to those of 5. However, in its IR spectrum no C-C double bond absorption was observed, and its $^1\text{H NMR}$ spectrum contained no signals for olefinic protons. These data were in agreement with a structure related to dihydrojuvabione. The presence in the $^1\text{H NMR}$ spectrum of a multiplet at $\delta 2.64$ showed the equatorial disposition of the geminal proton to the carboxylic group and suggested that the substitution in the cyclohexane ring was *cis*. The $^{13}\text{C NMR}$ spectrum confirmed these deductions, but it did not allow us to assign the configuration at C-1'. Hydrogenation of epijuabione (1) yielded the corresponding *trans* and *cis*-dihydroepijuabione isomers (9 and 2). One of which was identical to ester 2. This result, together with the fact that all sesquiterpenoids related to juvabione isolated in *A. pinsapo* belonged to the *epi* series, permitted us to propose an *S* configuration at C-1'. Consequently 6 has the structure of *cis*-dihydroepitodomatuiic acid.

Saponification of 3 gave a colourless oil, the mass spectrum of which showed the molecular ion at m/z 250 ($C_{15}H_{22}O_3$). In the IR spectrum, we observed bands at 1643 and 1617 cm^{-1} , assignable to C-C double bonds which were conjugated with carboxyl and ketone carbonyl groups, respectively. The $^1\text{H NMR}$ spectrum showed two doublets at $\delta 1.90$ and 2.15 due to methyl groups bound to a double bond and with *trans* and *cis* disposition with respect to a ketone group, as well as two multiplets of olefinic protons at $\delta 6.07$ and 7.12 . These data were consistent with a structure related to dehydrojuvabione. The $^{13}\text{C NMR}$ spectrum was in accordance with this proposal, the δ values of C-3 (28.7) and C-5 (26.1) being quite remarkable. This and the optical rotation value allowed us to establish the stereochemistry of the molecule, and to assign it the structure of 4'-dehydroepitodomatuiic acid (7).

Saponification of the most polar ester (4) gave a hydroxyacid with the molecular formula $C_{15}H_{26}O_3$ ($[M-H_2O]^{+}$, m/z 236). Its IR spectrum showed the presence of an alcoholic hydroxyl group (3400 cm^{-1}), a carboxyl

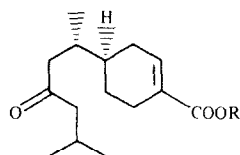
Table 1 ^1H NMR spectral data of compounds 5–9*

H	5	6	7	8	9
1	—	2.64m	—	—	—
2	7.10m	—	7.12m	7.10m	—
3'	—	—	—	3.75m	—
4'	—	—	6.07m	—	—
6'	0.86 d (6.4)†	0.88 d† (6.4)	2.15 d (1.0)	0.88 d (6.0)	0.88 d† (6.8)
7'	0.86 d (6.4)	0.89 d† (6.4)	1.90 d (1.0)	0.88 d (6.0)	0.89 d† (6.8)
8'	0.84 d (6.0)	0.82 d (6.0)	0.90 d (6.5)	0.88 d (6.0)	0.81 d (6.0)
OMe	—	—	—	—	3.64 s

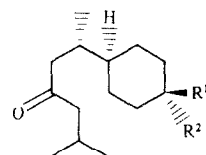
*5 and 9 300 MHz; 6–8 80 MHz (CDCl_3 , TMS as int. standard)

† Interchangeable value

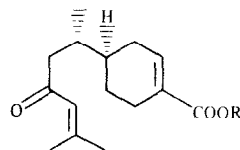
‡ Coupling constants (J in Hz) are given in parentheses



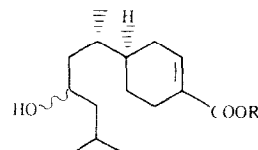
1 R = Me
5 R = H



2 R¹ = COOMe R² = H
6 R¹ = COOH R² = H
9 R¹ = H R² = COOMe



3 R = Me
7 R = H



4 R = Me 3'S
8 R = H 3'S
10 R = Me 3'R

group (1690 cm^{-1}) and a double bond conjugated with the former (1646 cm^{-1}). In the ^1H NMR spectrum, overlapping signals of methyl groups bound to methine and of an isopropyl group ($\delta 0.88$, d, 9H) were present, as were signals of a proton geminal to a hydroxyl group ($\delta 3.75$, m) and an olefinic proton β to a carboxyl group ($\delta 7.10$, m). In accordance with these data, the compound had a structure related to 3'-dihydrojuvabione (juvabiol). Again, the ^{13}C NMR data and the optical rotation value permitted us to define the stereochemistry. The values of $\delta 28.7$ and 26.0 for C-3 and C-5 respectively, were in agreement with those reported by Manville [13] for an *S* configuration at C-1' in epijuviabiol, whereas the ones corresponding to C-2' and C-4' ($\delta 42.4$ and 47.9) were consistent with an *S* configuration at C-3'. Therefore, the natural acid was assigned the structure of 3'-dihydroepitodomatuaic acid (8). In accordance with this proposal, sodium borohy-

dride reduction of epijuviabione (1) yielded a mixture of two substances, the ^{13}C NMR of which contained the signals of 4, and those Manville [13] had assigned to isoepijuviabiol (10).

EXPERIMENTAL

The wood was collected in June in Sierra Bermeja (Málaga, Spain) and was identified by Professor F. Valle (Department of Botany, University of Granada). The air-dried material (5.2 kg) was crushed and extracted with hexane in a Soxhlet apparatus for 12 hr. The extract (29.8 g) was defatted and, after dissolving in Et_2O , was extracted with NaOH solns (5, 2.5, and 10%) giving 10.85 g of a neutral fraction and 8.16 g of an acid one. The latter was treated with CH_3N_2 to yield the methyl esters which are easier to isolate and purify.

Table 2 ^{13}C NMR chemical shifts of compounds 1–10 (CDCl_3 , TMS as int. standard)*

C	1	2	3	4	5	6	7	8	9	10
1	130.2	39.5	130.2	130.1	130.2	39.5	129.9	129.9	43.3	130.1
2	139.2	27.1	139.5	139.7	141.6	27.0	141.9	142.3	27.1†	139.6
3	28.4	26.6†	28.5	28.3	28.8	26.9†	28.7	28.7	29.3‡	27.3
4	37.7	41.8	37.8	38.4	37.7	41.8	37.7	38.5	41.8	36.9
5	26.0	25.6†	26.2	25.9	26.1	25.6†	26.1	26.0	27.9‡	26.4
6	24.9	27.1	25.0	25.0	24.7	27.0	24.7	24.8	29.0†	25.0
7	167.7	175.6	168.0	167.9	172.8	181.3	172.8	172.5	176.3	167.8
8	51.4	51.4	51.5	51.9	—	—	—	—	51.4	51.3
1'	32.8	33.1	33.6	33.0	32.9	33.2	33.6	33.1	33.5	33.6
2'	47.9	48.0	49.0	42.3	48.0	48.1	49.0	42.4	48.0	42.7
3'	210.3	210.9	201.0	67.5	210.7	211.5	200.9	67.7	210.7	68.1
4'	52.3	52.3	124.1	47.8	52.5	52.4	124.1	47.9	52.4	46.9
5'	24.5	24.5	155.3	24.6	24.6	24.6	155.4	24.8	24.5	24.4
6'	22.5	22.5	20.7	22.2†	22.6†	22.7‡	20.8	23.4	22.5	21.8†
7'	22.5	22.5	27.7	23.2†	22.7†	22.6†	27.7	22.3	22.5	23.5†
8'	16.3	16.8	16.4	15.5	16.5	16.9	16.4	15.6	16.5	16.0

*1–4, 7 20 MHz; 5, 6, 8–10 75 MHz

†, ‡ Interchangeable values

The reaction mixture was chromatographed under pressure on a silica gel column eluted with hexane-Et₂O mixtures. Further chromatography on silica gel columns yielded the pure esters: **1** (170 mg), **2** (330 mg), **3** (300 mg) and **4** (84 mg).

Epijuvabione (**1**) Eluted with hexane-Et₂O (30:1), $[\alpha]_D^{25} + 60.0^\circ$ (CHCl₃; c 1.18)

cis-Dihydroepijuvabione (**2**). Eluted with hexane-Et₂O (49:1), $[\alpha]_D^{25} 0.0^\circ$ (CHCl₃; c 1.02).

4'-Dehydroepijuvabione (**3**). Eluted with hexane-Et₂O (97:3). AgNO₃-silica gel (1:4) gave pure **3**, $[\alpha]_D^{25} + 84.0^\circ$ (CHCl₃; c 0.51).

Epijuvabiol (**4**) Eluted with hexane-Et₂O (17:3). AgNO₃-silica gel (1:4) gave **4**, $[\alpha]_D^{25} + 50.2^\circ$ (CHCl₃; c 0.77). Saponification [MeOH-KOH, room temp., 8 (3) i.r. 21 hr] of the above esters yielded the corresponding natural acids. A shorter reaction period was used for **3** because longer periods led to retroaldolization. The reaction mixture was processed in the usual way.

Epitodomatuc acid (**5**). Colourless oil. $[\alpha]_D^{25} + 71.2^\circ$ (CHCl₃; c 1.07), IR $\nu_{\max}^{\text{film}} \text{ cm}^{-1}$: 3300-2500, 1710, 1686, 1646, 1462, 1423, 1366, 1273, 1085, 1030, 945, 739, MS (probe) 70 eV, m/z (rel. int.): 234 [M-H₂O]⁺ (15), 177 (12), 153 (20), 152 (57), 134 (100), 127 (36), 107 (44), 85 (41), 79 (40), 57 (56); UV $\lambda_{\max}^{\text{MeOH}} \text{ nm}$ (log ϵ) 221 (3.4)

cis-Dihydroepitodomatuc acid (**6**). Syrup. $[\alpha]_D^{25} 0.0^\circ$ (CHCl₃; c 1.19); IR $\nu_{\max}^{\text{film}} \text{ cm}^{-1}$: 3500-2500, 1706, 1452, 1371, 1238, 1181, 1136, 958, 842, MS (probe) 70 eV, m/z (rel. int.): 254 [M]⁺ (1), 154 (66), 136 (76), 127 (84), 109 (78), 108 (74), 85 (85), 57 (100)

4'-Dehydroepitodomatuc acid (**7**). Solid Mp (uncorr.) 56-58°, $[\alpha]_D^{25} + 94.0^\circ$ (CHCl₃; c 0.71), IR $\nu_{\max}^{\text{film}} \text{ cm}^{-1}$: 3700-2400, 1681, 1643, 1617, 1427, 1378, 1272, 1211, 1126, 1075, 1032, MS (probe) 70 eV, m/z (rel. int.): 251 [M+1]⁺ (1), 223 (1), 152 (2), 149 (8), 120 (3), 87 (10), 85 (61), 83 (100), 47 (17), 44 (13); UV $\lambda_{\max}^{\text{MeOH}} \text{ nm}$ (log ϵ) 246 (4.2)

3'-Dihydroepitodomatuc acid (**8**). Syrup. $[\alpha]_D^{25} + 51.6^\circ$ (CHCl₃; c 0.97), IR $\nu_{\max}^{\text{film}} \text{ cm}^{-1}$: 3700-2200, 3400, 1690, 1646, 1462, 1423, 1264, 1141, 949, 760, 701, MS (probe) 70 eV, m/z (rel. int.): 236 [M-H₂O]⁺ (8), 179 (20), 152 (19), 151 (15), 124 (100), 123 (42), 109 (11), 107 (19), 83 (10), 79 (11), 69 (19), 57 (8), UV $\lambda_{\max}^{\text{MeOH}} \text{ nm}$ (log ϵ) 219 (3.5)

Hydrogenation of epijuvabione (**1**) 297 mg of **1** was dissolved in MeOH (20 ml) containing Pd/C catalyst (10% Pd, 300 mg). Hydrogenation was performed at room temp at 1.6 atm for 4 hr. After removing the catalyst and solvent the crude product (250 mg) chromatographed on a silica gel column eluted with

hexane-Et₂O (24:1). Two products, (**2** and **9**) in roughly equal amounts, were isolated.

Reduction of epijuvabione (**1**) with NaBH₄ Compound **1** (1.2 mmol) in MeOH (5 ml) was treated with NaBH₄ (1.2 mmol) at room temp for 24 hr. The soln was neutralized with 2 M HCl, the MeOH evapd and the residue extracted with Et₂O (3 × 50 ml). The extracts were washed (3 × 50 ml), dried over Na₂SO₄ and then evapd to give a mixture (256 mg) which was chromatographed on a silica gel column eluted with hexane-Et₂O mixtures (4:1 and 3:1). 190 mg of a mixture of epijuvabiol (**4**) and isoepijuvabiol (**10**) was obtained. ¹³C NMR, Table 2.

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